

# Experimental Investigation of Nonideality and Nonadiabatic Effects under High Pressure Releases

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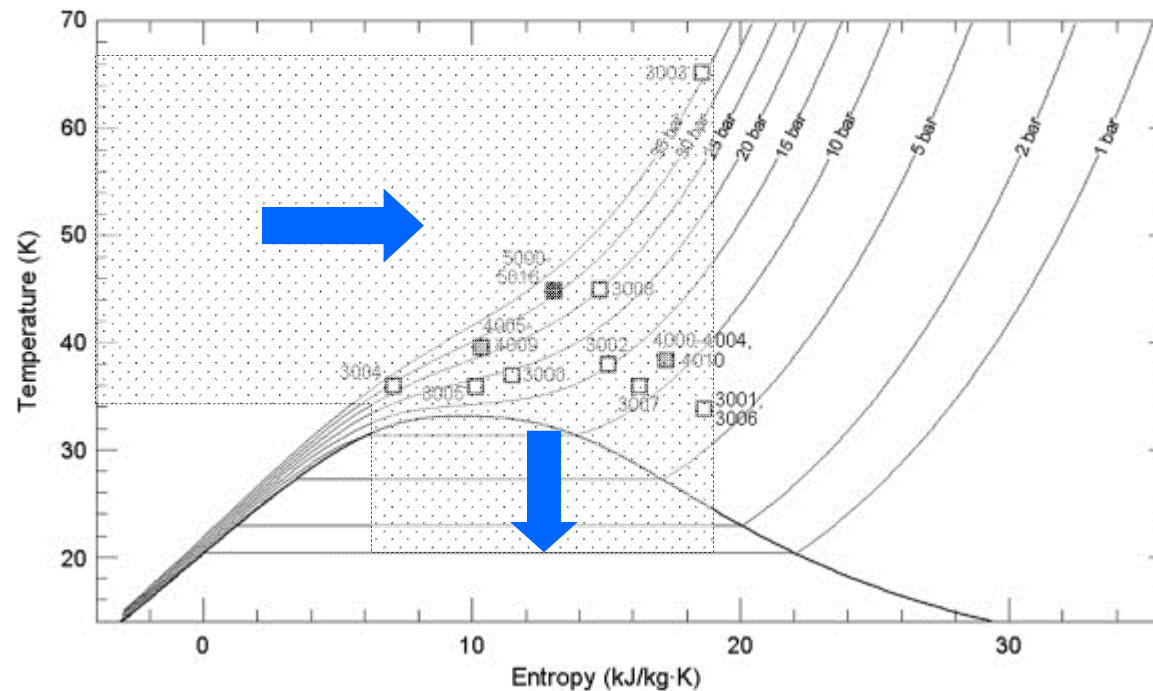
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# Background

The problem is related to the high pressure hydrogen releases in order to evaluate proper dynamics of the discharge and characteristic discharge time

## Temperature – entropy (T-S) – diagram of state of real para-hydrogen (NIST+icefuel data)



- Problems to be solved: nonideality of the blow-down process (two-phase flow); non adiabatic character of the discharge process

# Blow down mass flow rate

Solution of Bernoulli equation (integral form):

$$\dot{m} = C_D \cdot A_2 \cdot \max \left\{ \rho_2 \left[ -2 \left( \frac{p_1}{\rho_1} \right)^{\frac{p_2}{p_1}} \int_1^{\frac{p_2}{p_1}} \left( \frac{\rho_2}{\rho_1} \right)^{-1} d \left( \frac{p_2}{p_1} \right) \right]^{\frac{1}{2}} \right\}$$

$C_D = \frac{\dot{m}_{\text{exp}}}{\dot{m}_{\text{theor}}}$

**1, u – upstream conditions**  
**2, d – downstream conditions**

$$\dot{m} = C_D A_2 \left\{ \begin{array}{l} \left\{ \gamma_u p_u \rho_u \left( \frac{2}{\gamma_u + 1} \right)^{\frac{\gamma_u + 1}{\gamma_u - 1}} \right\}^{\frac{1}{2}} ; \quad \frac{p_d}{p_u} \leq \left( \frac{2}{\gamma_u + 1} \right)^{\frac{\gamma_u}{\gamma_u - 1}} \quad \text{critical (choked) flow} \\ \left\{ p_u \rho_u \frac{2\gamma_u}{\gamma_u - 1} \left[ \left( \frac{p_d}{p_u} \right)^{\frac{2}{\gamma_u}} - \left( \frac{p_d}{p_u} \right)^{\frac{\gamma_u + 1}{\gamma_u}} \right] \right\}^{\frac{1}{2}} ; \quad \frac{p_d}{p_u} > \left( \frac{2}{\gamma_u + 1} \right)^{\frac{\gamma_u}{\gamma_u - 1}} \quad \text{subcritical flow} \end{array} \right.$$

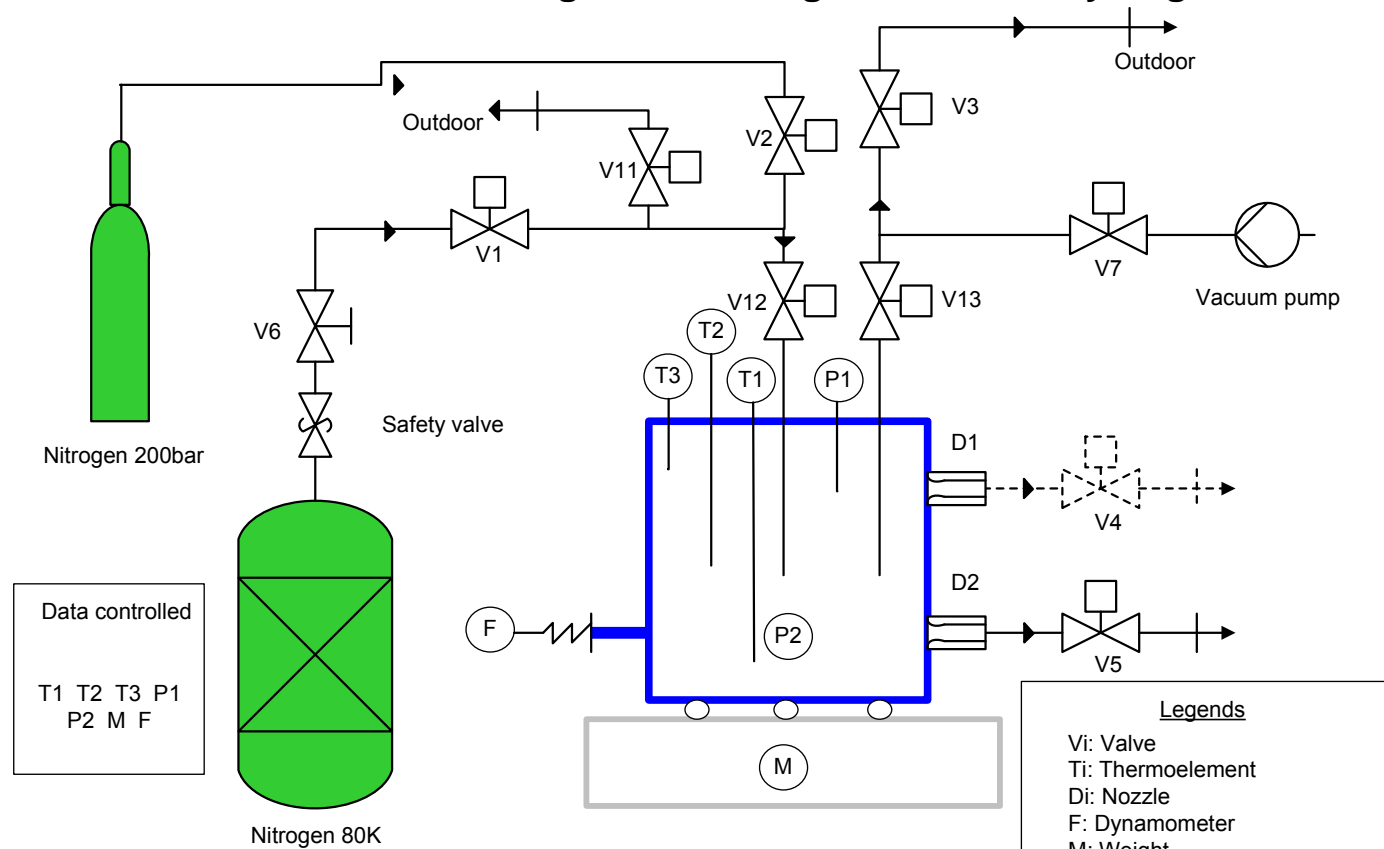
- The idea is to properly calculate the blow down mass flow rate for high pressure releases
- Validation against experiments to evaluate the  $C_D$  taking into account nonideality (two phase) and non adiabatic character of blow down process

# Objectives

- The objective of current work is to obtain detailed experimental data on the high-pressure releases in wide range of initial pressures and nozzle diameters to take into account nonideality of the process.
- In order to simplify the conditions for two-phase flow and for safety reasons, nitrogen will be used instead of the pressurized hydrogen.
- With this work, a capability of numerical and theoretical models for high pressure hydrogen releases will be validated against time-dependent experimental data

# Experimental facility

A gaseous system of DISCHA facility for transient two-phase blow-down tests with gaseous nitrogen instead of hydrogen

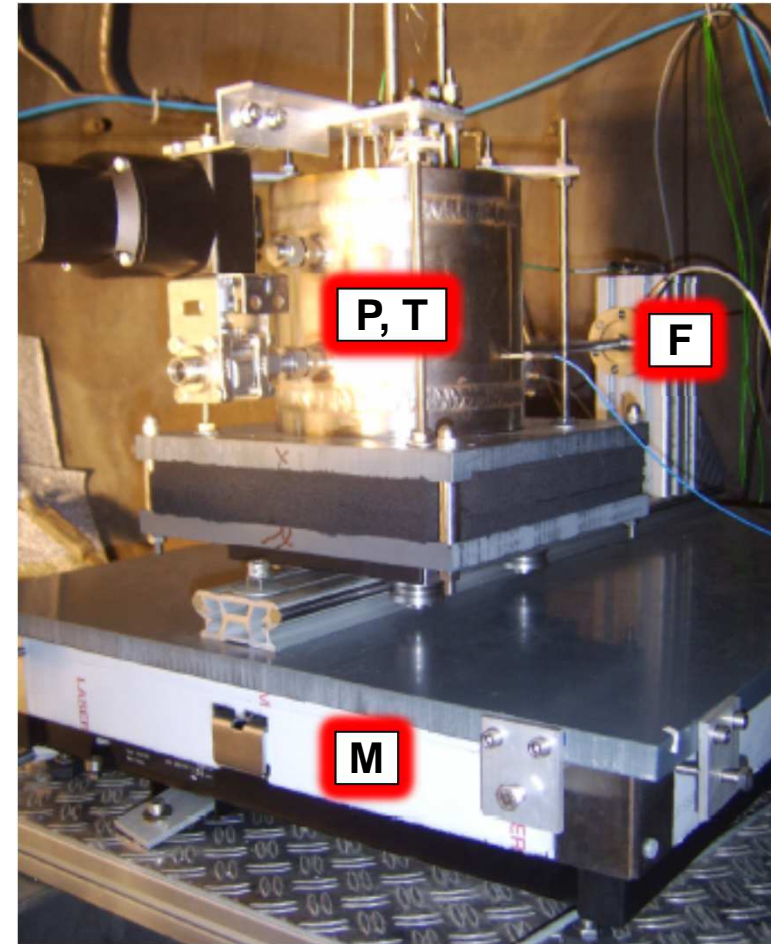
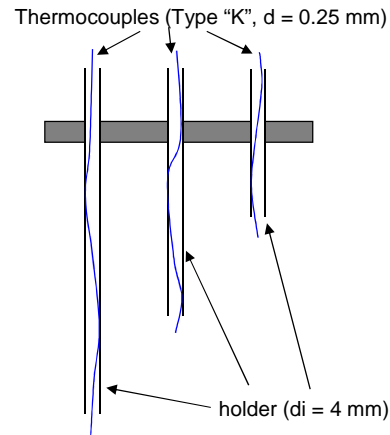


- Size of internal volume: 2.81 dm<sup>3</sup>
- Initial pressure: 5 ... 200 bar
- Initial temperature: 300K
- Two nozzle positions: D1, D2
- Nozzle diameters: 0.5, 1, 2, 3, 4 mm
- 2 piezo-resistive pressure transducers (P1, P2)
- 3 thermocouples (T1-T3)
- 1 force transducer (F)
- 1 scales (M)

**Legends**

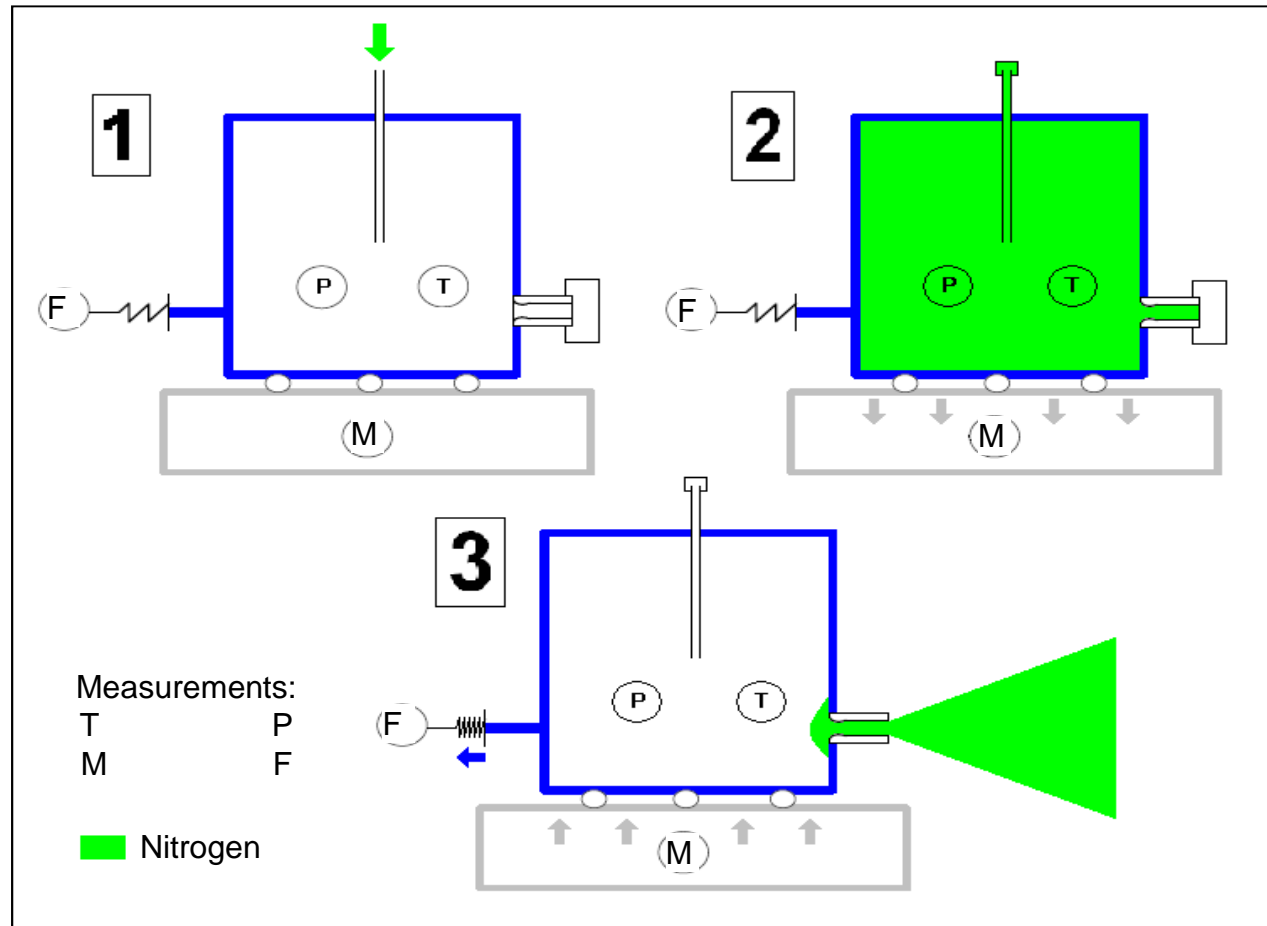
- Vi: Valve
- Ti: Thermoelement
- Di: Nozzle
- F: Dynamometer
- M: Weight
- Pi: Pressure sensor

# Experimental facility (side view)



- |                            |                      |   |
|----------------------------|----------------------|---|
| ■ Size of internal volume: | 2.81 dm <sup>3</sup> | ■ 2 piezo-resistive pressure transducers (P1, P2) |
| ■ Initial pressure:        | 5 ... 200 bar        | ■ 3 thermocouples (T1-T3)                         |
| ■ Initial temperature:     | 300K                 | ■ 1 force transducer (F)                          |
| ■ Two nozzle positions:    | D1, D2               | ■ 1 scales (M)                                    |
| ■ Nozzle diameters:        | 0.5, 1, 2, 3, 4 mm   |   |

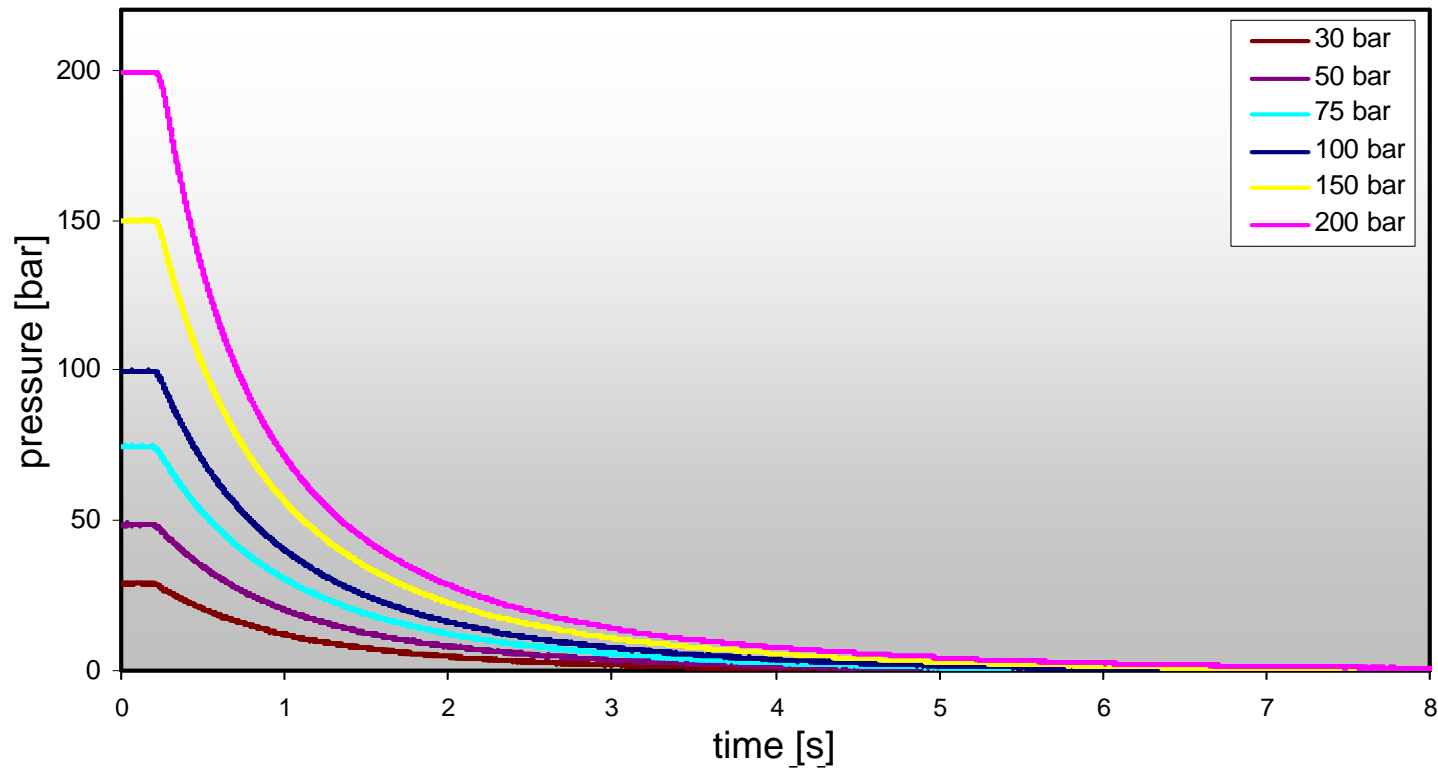
# Test procedure



- 1 Pre-evacuation and gas filling
- 2 Equilibrium of state (P, T)
- 3 Blow down process(T, P, F, M)

■ Simultaneous temperature, pressure, force and weight measurements provide independent measurements of mass flow rate

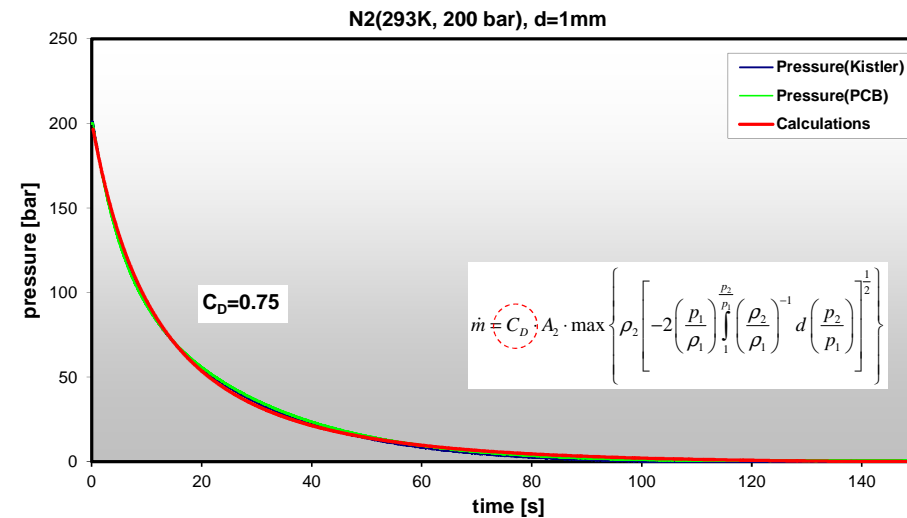
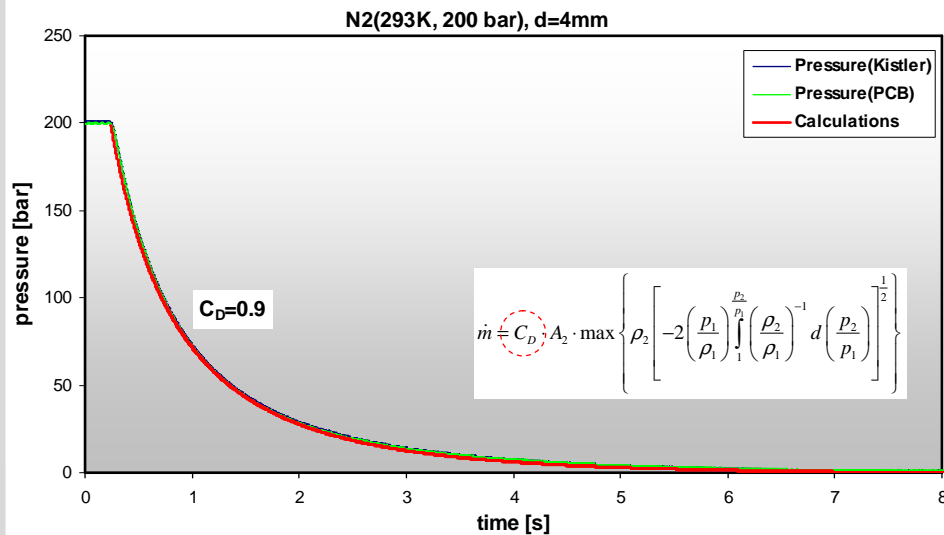
# Experimental results: pressure dynamics



- Very good reproducibility of the pressure behavior
- Characteristic pressure discharge time changes from 8 sec for 4-mm nozzle to 600 sec for 0.5-mm nozzle almost independent of initial pressure

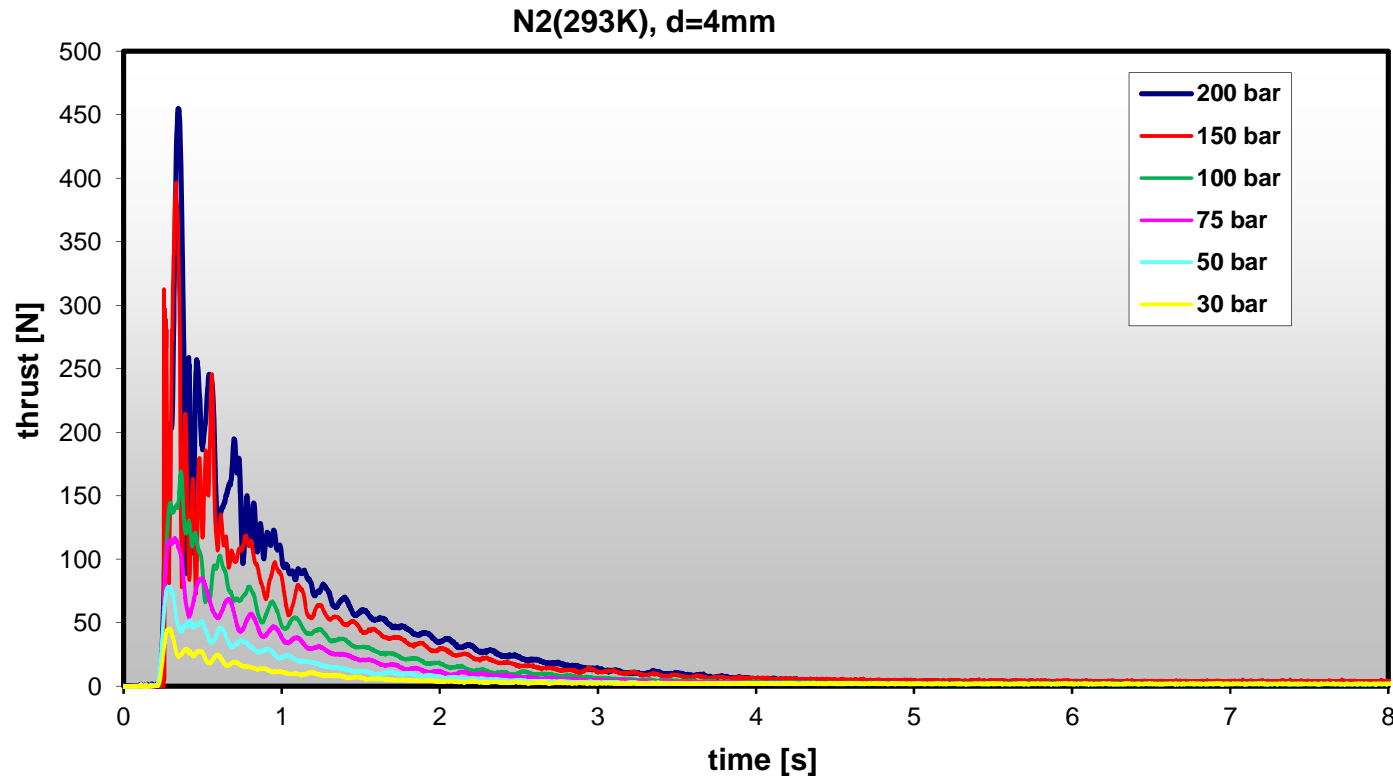


# Calculations: a comparison with pressure measurements



- Very good agreement of experimental data and theoretical calculations
- The discharge coefficient changes from 0.9 to 0.75 with nozzle diameter decrease from 4 to 1 mm inner diameter

# Experimental results: thrust measurements

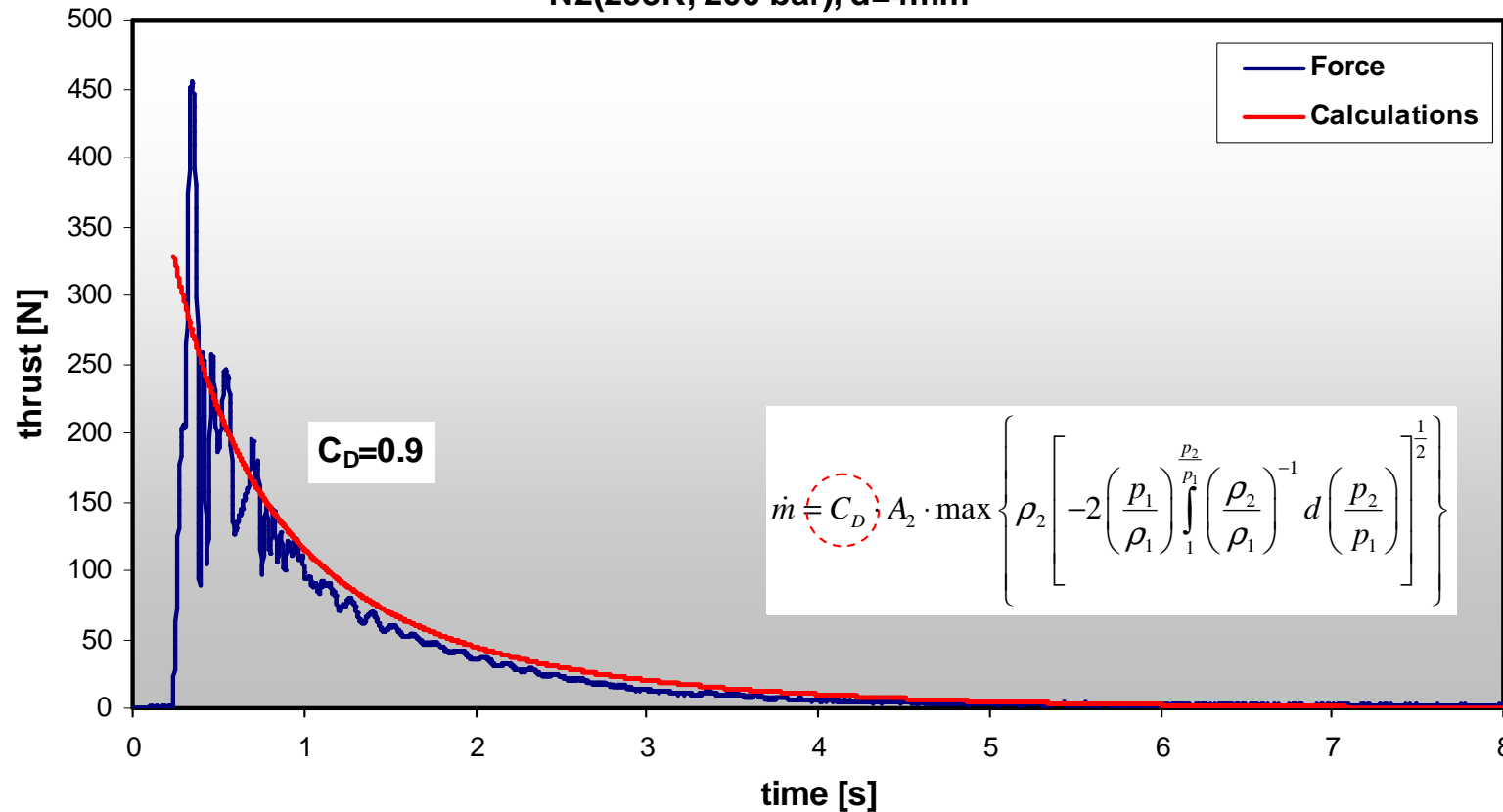


- Experimental dependencies of thrust data vs. time behave according to the initial pressure
- Some oscillating process was observed due to mechanical vibration of the system

# Calculations: a comparison with thrust measurements

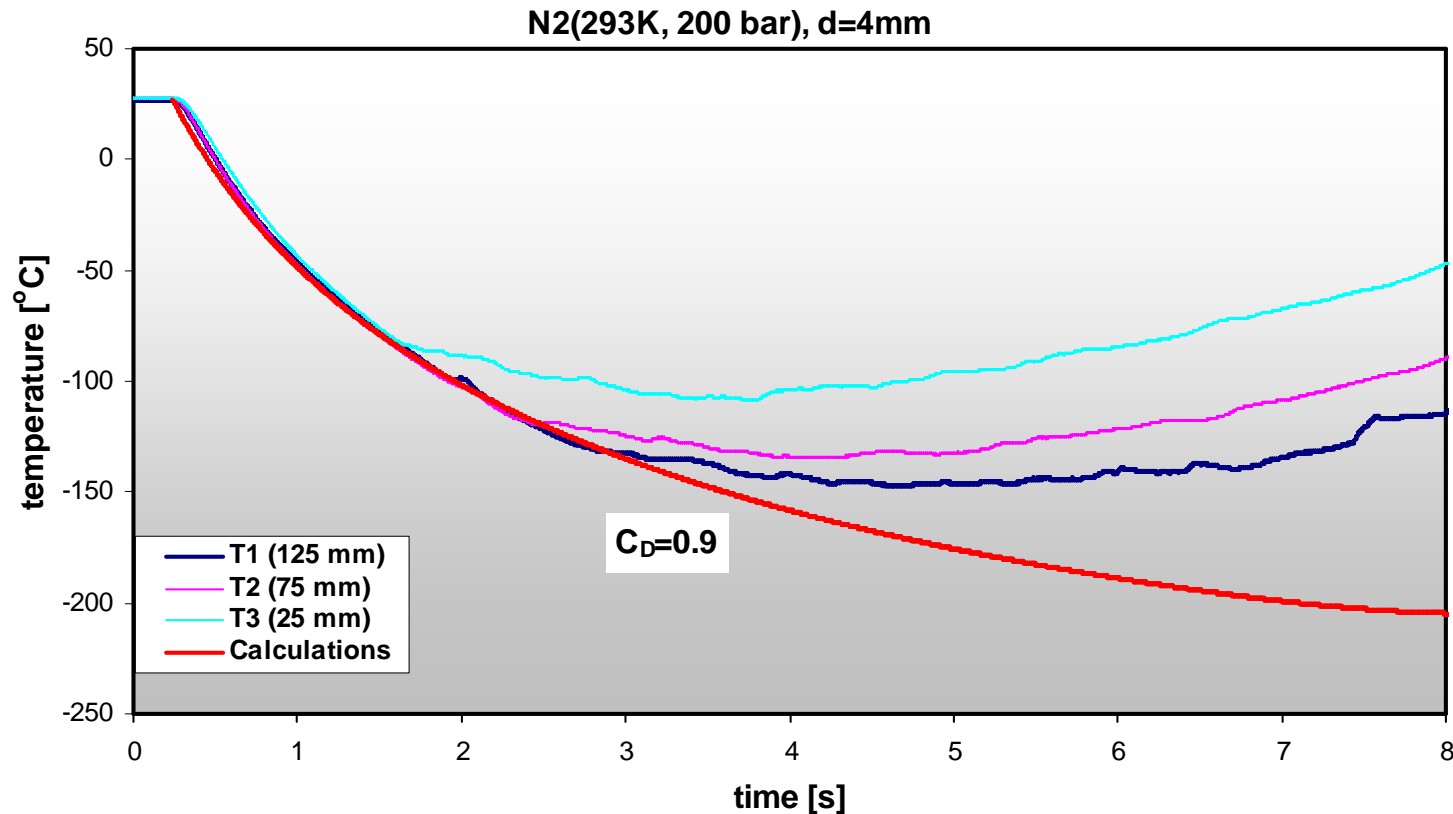
$$F = \dot{m}V_e + (p_e + p_0)A_e$$

N2(293K, 200 bar), d=4mm



- Experimental data on thrust measurements fit very well with theoretical calculations using the same discharge coefficient as for pressure dynamics

# A comparison of experimental temperature measurements and calculations

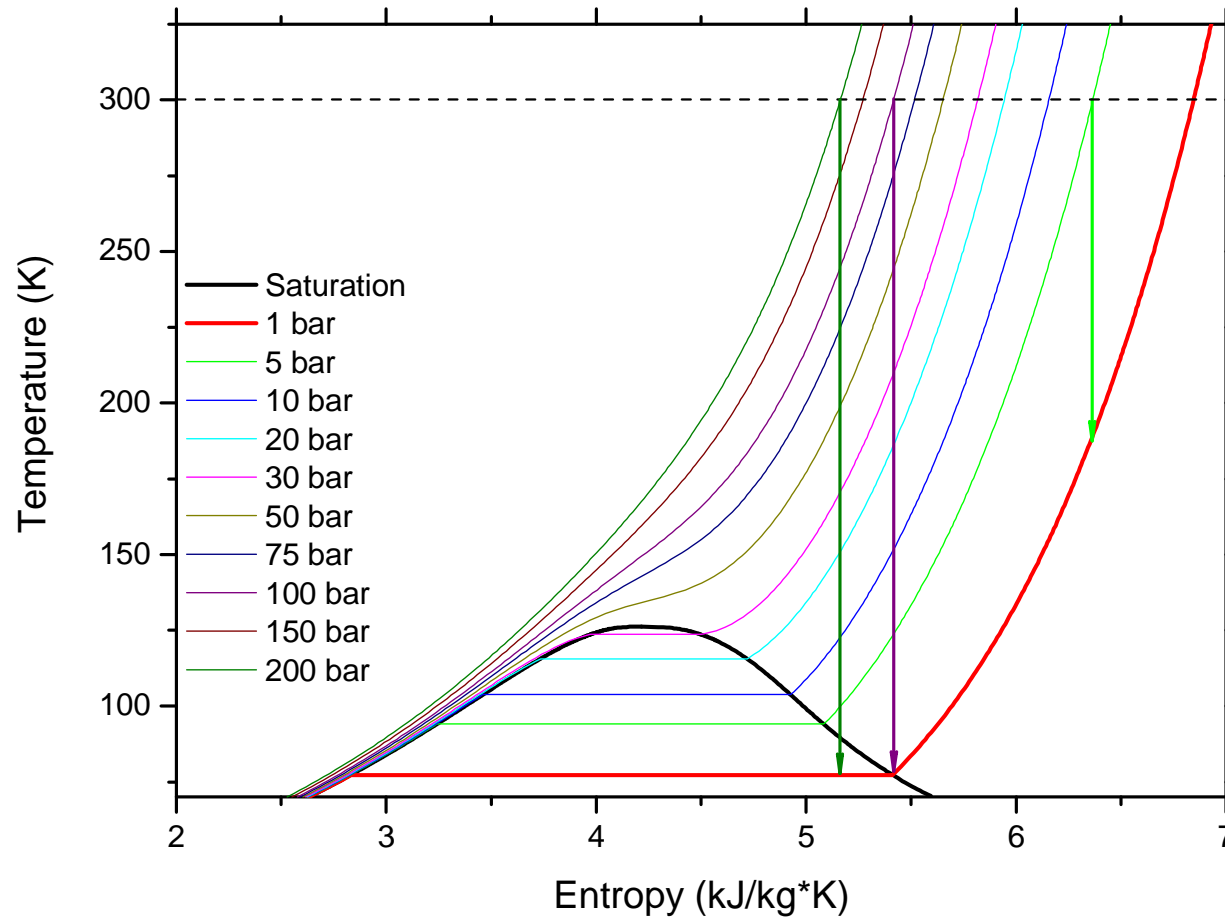


- The biggest deviation of theoretical and experimental values was found for temperature measurements
- Main reason was the nonadiabatic process due to heat exchange gas – solid walls
- The longer was the blow-down process, the higher deviation occurred

# Experimental data analysis

## Temperature – entropy (T-S) – diagram of state of real nitrogen (NIST)

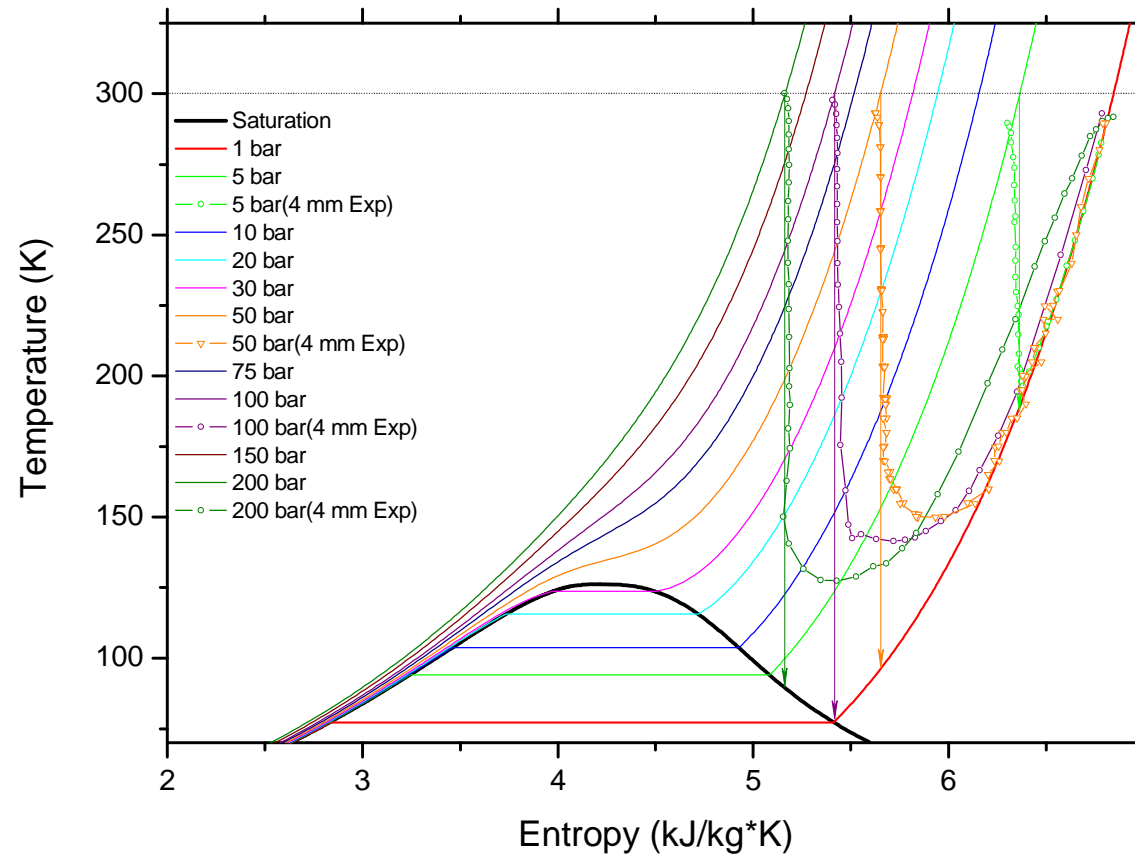
NIST Nitrogen Equation of State



- At initial pressure above 100 bar two-phase flow may occur

# Experimental data analysis

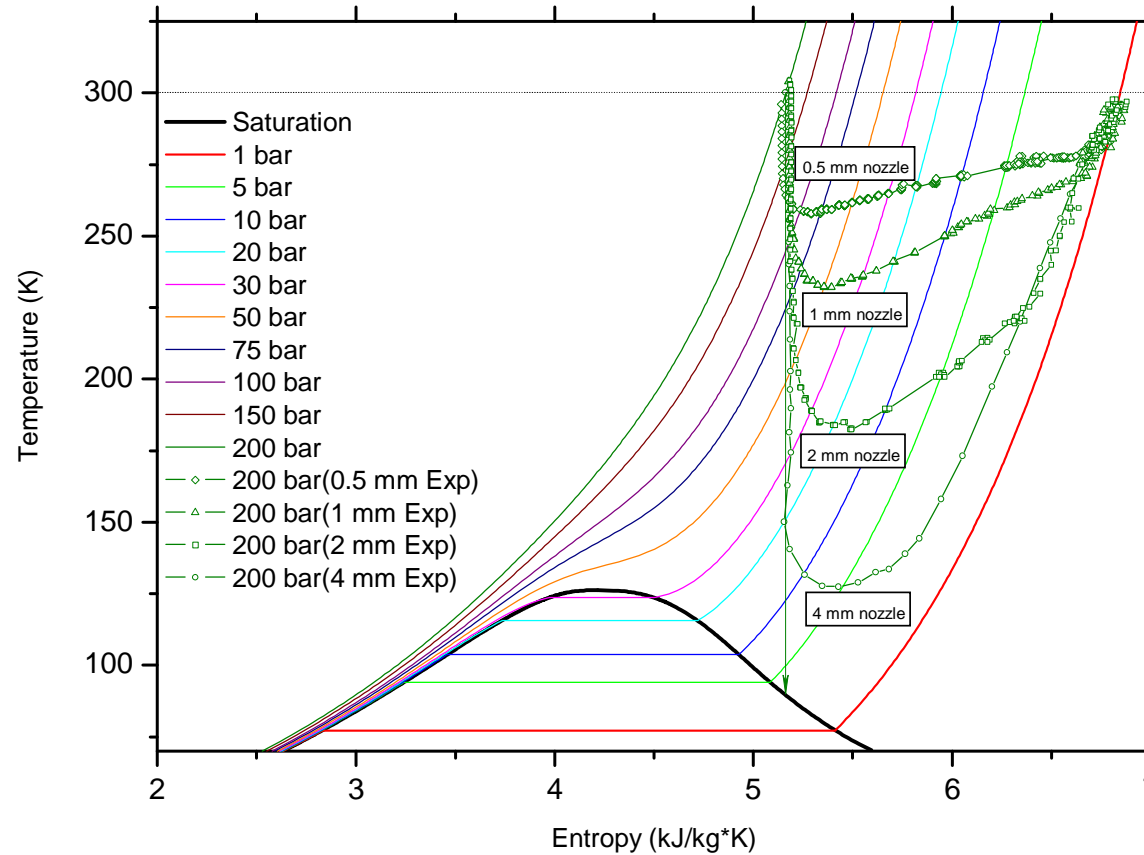
## Real nitrogen release at different initial pressures (4-mm nozzle)



- For 4-mm nozzle the entropy deviation appears when temperature difference reaches 120 – 150K due to heat transfer gas – solid wall
- Non- adiabatic blow down process occurs approaching subcritical blow down regime
- This was the reason why we did not reach the two-phase blow down process

# Experimental data analysis

## Real nitrogen release at 200 bar and different nozzle diameter



- The less nozzle diameter and the longer the blow down process, the lower the temperature when non adiabatic effect or entropy deviation appears (at 200 bar):

0.5-mm nozzle  $\Delta T = 40K$  ; 1-mm nozzle  $\Delta T = 60K$  ;

2-mm nozzle  $\Delta T = 120K$  ; 4-mm nozzle  $\Delta T = 170K$  ;

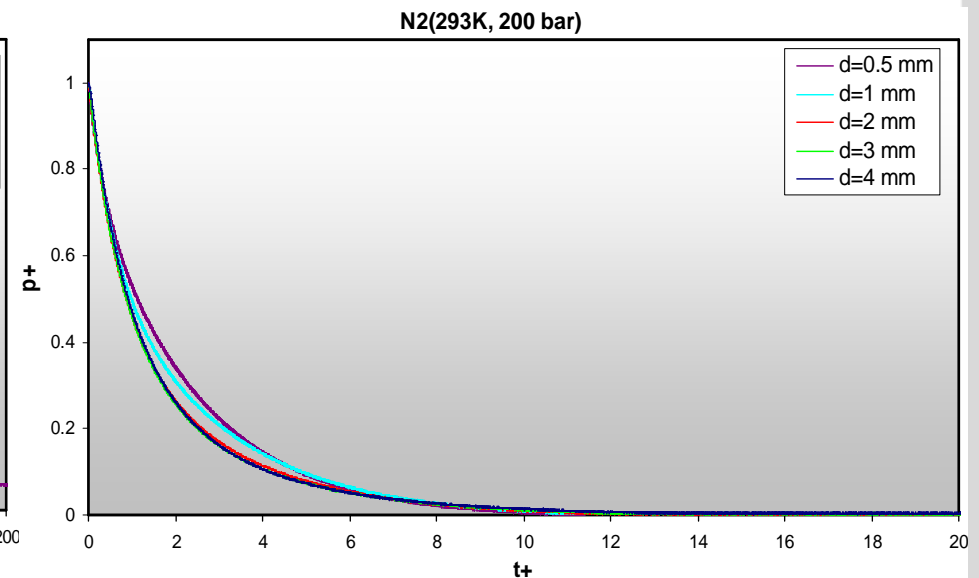
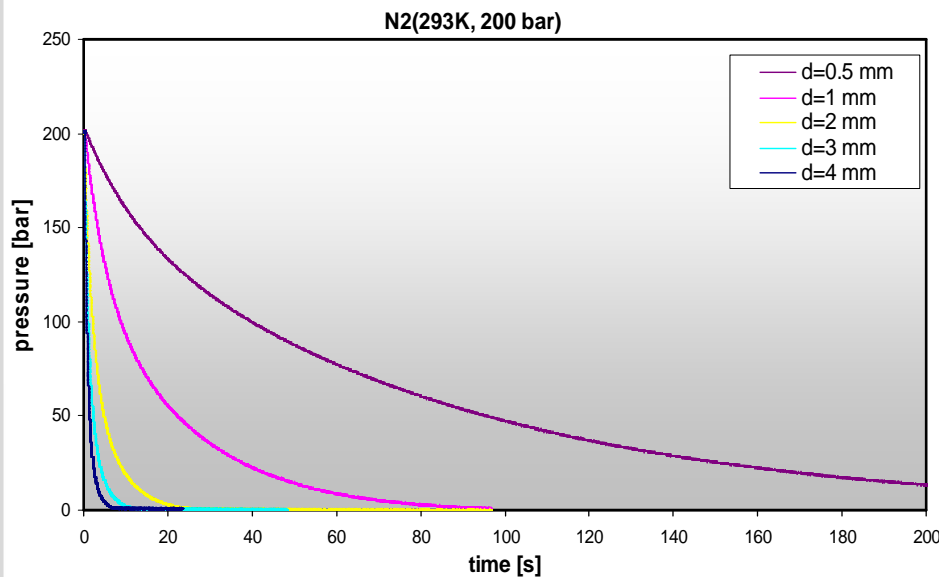
# Scaling of transient discharge pressures

$$p^+ = \left[ 1 + \left( \frac{\gamma-1}{2} \right) \left( \frac{\gamma+1}{2} \right)^{\frac{-(\gamma+1)}{2(\gamma-1)}} \cdot t^+ \right]^{\frac{-2\gamma}{\gamma-1}}$$

$p^+ = p(t)/p_0$  - dimensionless pressure;

$t^+ = t/t_{char}$  - characteristic release time;

$t_{char} = V/(A \cdot c_0)$  - characteristic release time



- Scaling by dimensionless  $p^+$  and  $t^+$  results in very good agreement of the tests with different initial pressures for the nozzle diameter more than 2 mm.
- There is some difference appears for smallest nozzle diameters due to the above discussed heat transfer effects and the discharge time. The slowest experiments (0.5 and 1 mm nozzles) show the highest values for  $p^+(t^+)$ .



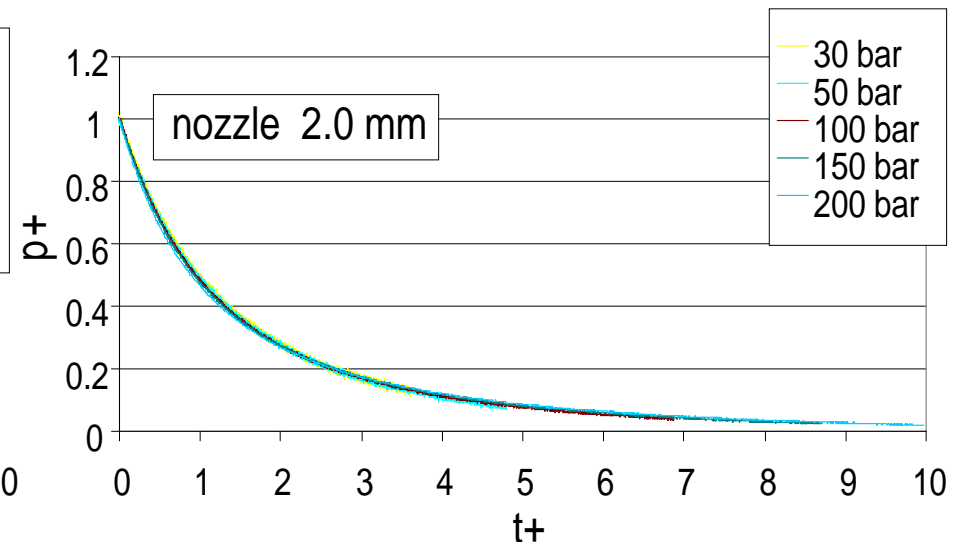
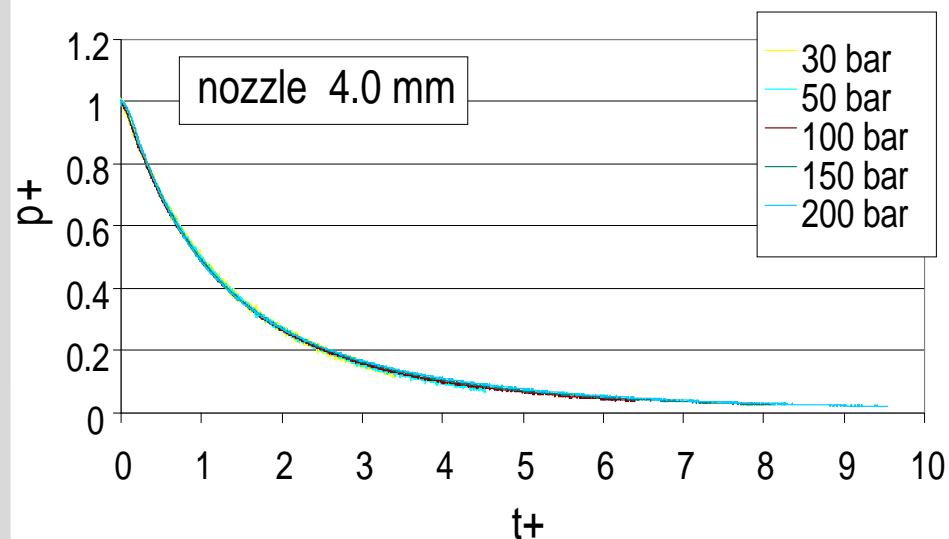
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$p^+ = p(t)/p_0$  - dimensionless pressure;

$t^+ = t/t_{\text{char}}$  - characteristic release time;

$t_{\text{char}} = V/(A \cdot c_0)$  - characteristic release time



- Characteristic time  $t^+$  includes the sound speed of the gas  $c_0$  in its initial state  $p_0/T_0$ , which varies significantly with the initial pressure.
- Independent of that the scaling equation was originally derived for ideal gases with constant  $\gamma$  and constant sound speed  $c_0$  during the blow-down process it allows a very good scaling of the present non-ideal high-pressure discharge experiments with nitrogen
- The measured pressures  $p^+(t^+)$  were scaled well from the initial pressure  $p_0$  down to  $p_{\text{end}} = 3$  bar to remain in the choked flow regime, but even further the difference is rather small

## Conclusions

- A small-scale facility for transient discharge of cryogenic nitrogen was fabricated and tested with gaseous nitrogen as an inert hydrogen substitute. Different orifice sizes (0.5, 1, 2, 3, 4 mm) and initial N<sub>2</sub> pressures (30 – 200 bar) were investigated.
- The measured time-dependent data for vessel discharge pressure, thrust, discharge mass flow, and gas temperatures could be well reproduced using the NIST database for the real gas equation-of-state of nitrogen. This verification for nitrogen also assures the EOS for hydrogen, which is based on the same methodology.
- The newly developed critical discharge analysis method for a pure substance predicts correctly the transient blow-down of a high-pressure gas system. The measured pressure histories could be scaled very well using initial pressure and sound speed of the gas, vessel volume and nozzle area as characteristic quantities.
- New results about the heat transfer effects in blow-down of gaseous high-pressure systems have been obtained. For relatively small nozzle diameter and lower initial pressure the heat from surrounding may completely eliminate the two-phase scenario of high pressure release.
- The facility is ready for extension of the experiments to liquid nitrogen and investigation of cryogenic two-phase discharge in a future phase